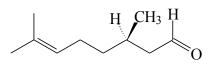
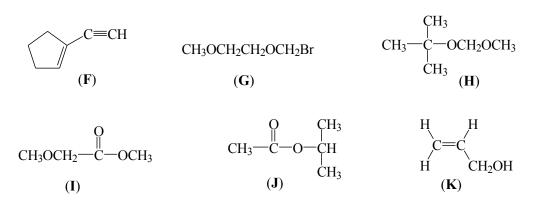
CHEM1611 Problem Sheet 8 (Week 10)

Work through the ChemCAL modules "Nucleophilic Addition to Carbonyl Groups" and "Organic Acids and Bases".

1. (+)-Citronellal is a widely occurring natural product, present in citronella oil, lemon and lemon grass. It is used as a soap perfume and in insect repellents.



- (a) Give the molecular formula for citronellal.
- (b) List the functional groups present in citronellal.
- (c) Give the constitutional formula of the major product(s) formed when citronellal is treated with each of the following reagents.
 - (i) $Cr_2O_7^{2-}/H^+$
 - (ii) excess CH_3OH / catalytic amount H_2SO_4
 - (iii) NaBH₄ in CH₃OH followed by H^+/H_2O
 - (iv) H_2/Pd in ethanol
 - (v) $3 \text{ M H}_2\text{SO}_4$
 - (vi) HCl in CCl₄ solvent
- 2. Predict the products for the following reactions.
 - (a) acetone heated in excess methanol with a catalytic amount of H_2SO_4
 - (b) ethylammonium chloride with dilute sodium hydroxide
- 3. Consider the compounds F, G, H, I, J and K.

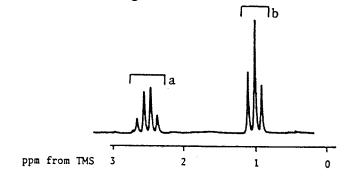


- (a) Which of the compounds strongly absorb IR in the range 1650-1800 cm $^{-1}$?
- (b) Which of the compounds strongly absorb IR in the range $3200-3700 \text{ cm}^{-1}$?
- (c) Which of the compounds strongly absorb UV radiation?
- (d) Which of the compounds have a ¹H NMR spectrum consisting of only three singlets?
- (e) Which of the compounds will show peaks of approximately equal intensity at m/z 168 and 170 in its mass spectrum?

4. Consider the compounds having constitutional formulas **H** to **M**.

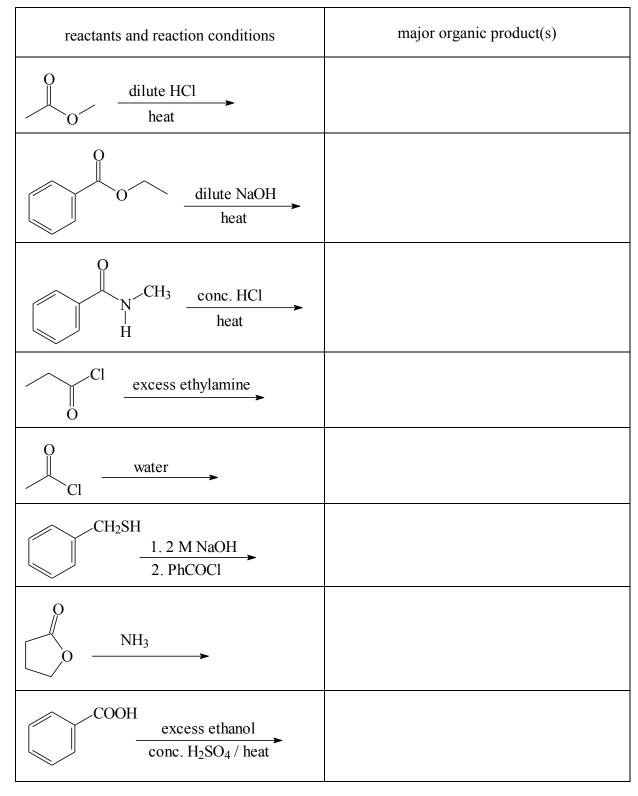
$$\begin{array}{cccc} CH_3 - CH_2 - CH_2 - C - H & CH_3 - CH - CH_2 - C - H \\ (H) & CH_3 & O \\ (I) & (J) \end{array}$$

- (a) Which of the compounds **H** to **M** would have a parent ion in the mass spectrum at m/z 86?
- (b) Which one of the compounds **H** to **M** has the ¹H NMR spectrum below?
- (c) State the relative areas of signals a:b.



(d) Which (if any) of compounds **H** to **M** strongly absorb IR in the ranges 1650-1750, 2850-2950 and 3200-3600 cm⁻¹?

Problem sheet 8 continues on the next page.



5. Give the major organic product(s) formed in the following reactions.

6.	Fats and oils are triesters of glycerol.	СН ₂ ОН	
	What fat is prepared from stearic acid and	СНОН	CH ₃ (CH ₂) ₁₆ COOH
	what reaction conditions are required to convert it back to glycerol?	CH ₂ OH	stearic acid
		glycerol	